

PCTWORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶: A61K 47/48, C08G 65/32	A1	(11) International Publication Number: WO 99/45964 (43) International Publication Date: 16 September 1999 (16.09.99)
(21) International Application Number: PCT/US99/05333 (22) International Filing Date: 11 March 1999 (11.03.99) (30) Priority Data: 60/077,700 12 March 1998 (12.03.98) US (63) Related by Continuation (CON) or Continuation-in-Part (CIP) to Earlier Application US 60/077,700 (CON) Filed on 12 March 1998 (12.03.98) (71) Applicant (for all designated States except US): SHEAR-WATER POLYMERS, INCORPORATED [US/US]; 1112 Church Street, Huntsville, AL 35801 (US). (72) Inventors; and (75) Inventors/Applicants (for US only): HARRIS, J., Milton [US/US]; 3119 Highland Plaza, Huntsville, AL 35801 (US). KOZLOWSKI, Antoni [US/US]; 1500-13D Sparkman Drive, Huntsville, AL 35816 (US). (74) Agents: PEDIGO, Paul, F. et al.; Alston & Bird LLP, P.O. Drawer 34009, Charlotte, NC 28234-4009 (US).		(81) Designated States: AE, AL, AM, AT, AT (Utility model), AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, CZ (Utility model), DE, DE (Utility model), DK, DK (Utility model), EE, EE (Utility model), ES, FI, FI (Utility model), GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SK (Utility model), SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i>
(54) Title: POLY(ETHYLENE GLYCOL) DERIVATIVES WITH PROXIMAL REACTIVE GROUPS		
(57) Abstract An activated, substantially water-soluble poly(ethylene glycol) is provided having a linear or branched poly(ethylene glycol) backbone and at least one terminus linked to the backbone through a hydrolytically stable linkage, wherein the terminus is branched and has proximal reactive groups. The free reactive groups are capable of reacting with active moieties in a biologically active agent such as a protein or peptide thus forming conjugates between the activated poly(ethylene glycol) and the biologically active agent.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

POLY(ETHYLENE GLYCOL) DERIVATIVES
WITH PROXIMAL REACTIVE GROUPS

Cross-Reference to Related Applications

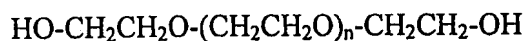
This application is related to copending Provisional Application
Serial No. 60/077,700, filed March 12, 1998, incorporated by reference in its
5 entirety, and claims the benefit of its filing date under 35 U.S.C. §119(e).

FIELD OF THE INVENTION

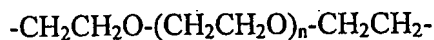
This invention relates to derivatives of polyethylene glycol and
related hydrophilic polymers suitable for chemical coupling to another molecule,
10 including, for example, proteins, enzymes, small drugs, and the like.

BACKGROUND OF THE INVENTION

Chemical attachment of the hydrophilic polymer poly(ethylene
glycol) ("PEG") to molecules and surfaces is of great utility in biotechnology. In
its most common form PEG is a linear polymer terminated at each end with
15 hydroxyl groups:



This polymer can be represented in brief form as HO-PEG-OH where it is
20 understood that the -PEG- symbol represents the following structural unit:



In typical form n ranges from about 10 to about 2000.

PEG is commonly used as methoxy PEG-OH, or mPEG in brief, in which one terminus is the relatively inert methoxy group, while the other terminus is a hydroxyl group that is subject to ready chemical modification.

5



PEG is also commonly used in branched forms that can be prepared by addition of ethylene oxide to various polyols, such as glycerol, pentaerythritol and sorbitol. For example, the four-arm, branched PEG prepared from
10 pentaerythritol is shown below:



The branched PEGs can be represented in general form as R(-PEG-
15 OH)_n in which R represents the central "core" molecule, such as glycerol or pentaerythritol, and n represents the number of arms.

Branched PEGs can also be prepared in which two PEG "arms" are attached to a central linking moiety having a single functional group capable of joining to other molecules; e.g., Matsushima *et al.*, (Chem. Lett., 773, 1980) have
20 coupled two PEGs to a central cyanuric chloride moiety.

PEG is a well known polymer having the properties of solubility in water and in many organic solvents, lack of toxicity, and lack of immunogenicity. One use of PEG is to covalently attach the polymer to insoluble molecules to make the resulting PEG-molecule "conjugate" soluble. For example, it has been shown
25 that the water-insoluble drug paclitaxel, when coupled to PEG, becomes water-soluble. Greenwald, *et al.*, *J. Org. Chem.*, 60:331-336 (1995).

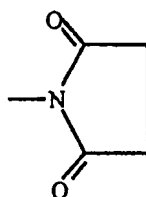
In related work, U.S. Patent 4,179,337 to Davis *et al.* discloses that proteins coupled to PEG have enhanced blood circulation lifetime because of reduced rate of kidney clearance and reduced immunogenicity. These and other
30 applications are also described in *Biomedical and Biotechnical Applications of Polyethylene Glycol Chemistry*, J. M. Harris, Ed., Plenum, New York (1992), and *Poly(ethylene glycol) Chemistry and Biological Applications*, J.M. Harris and S. Zalipsky, Eds., ACS, Washington DC (1997).

To couple PEG to a molecule such as a protein, it is often necessary to "activate" the PEG to prepare a derivative of the PEG having a functional group at the terminus. The functional group can react with certain moieties on the protein such as an amino group, thus forming a PEG-protein conjugate. Many activated
5 derivatives of PEG have been described. An example of such an activated derivative is the succinimidyl succinate "active ester":



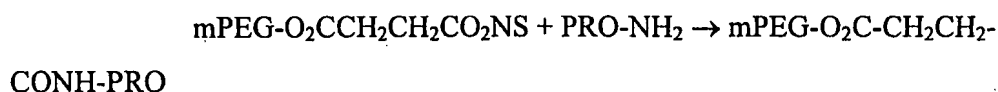
where NS =

10



Hereinafter, the succinimidyl active ester moiety will be represented as -CO₂-NS in chemical drawings.

The succinimidyl active ester is a useful compound because it reacts
15 rapidly with amino groups on proteins and other molecules to form an amide linkage (-CO-NH-). For example, U.S. Patent 4,179,337 to Davis *et al.* describes coupling of this derivative to proteins (represented as PRO-NH₂):



20

Bifunctional PEGs with active groups at both ends of the linear polymer chain are also useful compounds when formation of a crosslinked insoluble network is desired. Many such bifunctional PEGs are known in the art. For example, U.S. Patent 5,162,430 to Rhee, *et al.* discloses using such bifunctional PEGs to crosslink collagen.

25

Reactive PEGs have also been synthesized in which several active functional groups are placed along the backbone of the polymer. For example, lysine-PEG conjugates have been prepared in the art in which a number of activated groups are placed along the backbone of the polymer. Zalipsky *et al.* *Bioconjugate Chemistry*, 4:54-62 (1993).

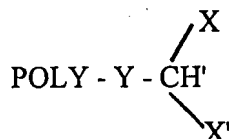
U.S. Patent 5,283,339 to Arnold *et al.* discloses PEG compounds capable of chelating metals. The PEG compounds have a terminal metal chelating group which has two free carboxylic acid or amino groups, typically linked to a nitrogen atom. The PEG compounds are used to extract and precipitate proteins from solutions with the carboxylic acid or amino groups together with the nitrogen atom capable of forming ionic complexes with metal ions. However, the metal chelating groups disclosed in the patent generally are not useful in covalently coupling the PEG compounds to proteins, peptides, or small drugs bearing functional groups such as amines. The patent does not teach forming an activated PEG derivative for covalently coupling to another molecule to form a conjugate.

SUMMARY OF THE INVENTION

The invention described herein provides a water soluble polymer such as poly(ethylene glycol) or related polymers that have a branched moiety at one end of the polymer chain and two free reactive groups linked to the branched moiety for covalent attachment to another molecule. Each reactive moiety can have a tethering group, such as an alkyl chain, linking a reactive group to the branched moiety. Thus, the branched terminus allows the activated water soluble polymer of this invention to react with two molecules to form conjugates.

Because a tethering group having a desirable length can be selected in preparing an activated polymer, the two reactive groups can be held at a predetermined distance apart from each other. The two molecules conjugated to the activated polymer through the two reactive groups can also be held at a predetermined distance apart. Accordingly, an activated PEG is provided in accordance with the invention having two free reactive moieties branching out from one PEG chain at a branched moiety. The two free reactive moieties are capable of reacting with biologically active agents such as proteins, thereby linking the activated polymer to the biologically active agents.

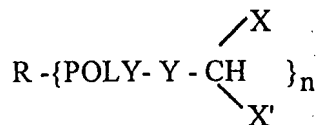
In accordance with one embodiment of this invention, an activated water soluble polymer is provided having the formula:



wherein POLY is a water soluble, substantially non-immunogenic polymer backbone, Y is a hydrolytically stable linkage, X and X' are reactive groups capable of reacting with a moiety in another molecule such as a protein. Typically, the polymer backbone is selected from the group consisting of linear and branched poly(ethylene glycol), linear and branched poly(alkylene oxide), linear and branched poly(vinyl pyrrolidone), linear and branched poly(vinyl alcohol), linear and branched polyoxazoline, linear and branched poly(acryloylmorpholine), and derivatives thereof. Preferably, the polymer backbone is poly(ethylene glycol) or a derivative thereof. The polymer backbone POLY can have a capping group selected from the group consisting of -OH, alkyls, and -Y-CHXX' wherein Y, X and X' are as described above and can be the same or different on each terminus of the PEG.

In a preferred embodiment, X and X' are represented by -W-Z and -W'-Z' respectively, in which Z and Z' represent reactive moieties for conjugating the polymer to another molecule. W and W' represent tethering groups comprising a substantially linear chain of atoms, e.g., alkyl chains, ether chains, ester chains, amide chains, and combinations thereof. Examples of the reactive moieties include, but are not limited to, active esters, active carbonates, aldehydes, isocyanates, isothiocyanates, epoxides, alcohols, maleimides, vinylsulfones, hydrazides, dithiopyridines, and iodoacetamides.

In another embodiment of the activated polymer of this invention, the activated water soluble polymer has the formula:



wherein

R is a central branch core;

POLY is a water soluble substantially non-immunogenic polymer;

Y is a hydrolytically stable linkage;

n is from 2 to 200;

5 X and X' are reactive groups capable of reacting with a moiety in another molecule such as a protein.

Many central branch core molecules for preparing branched or dendritic PEGs are known and can all be used for R. Typically, R can be a moiety derived from lysine, glycerol, pentaerythritol, or sorbitol. Suitable polymer
10 backbones include, but are not limited to, linear and branched poly(ethylene glycol), linear and branched poly(alkylene oxide), linear and branched poly(vinyl pyrrolidone), linear and branched poly(vinyl alcohol), linear and branched polyoxazoline, linear and branched poly(acryloylmorpholine), and derivatives thereof. Preferably, is poly(ethylene glycol) or a derivative thereof is used as the
15 polymer backbone.

The reactive groups X and X' can be reactive moieties directly linked to the branching moiety -CH. Preferably, X and X' have a tethering group and are represented by -W-Z and -W'-Z' respectively, in which Z and Z' represent reactive groups for conjugating the polymer to another molecule. W and W'
20 represent tethering groups comprising a substantially linear chain of atoms, e.g., alkyl chains, ether chains, ester chains, amide chains, and combination thereof. Examples of the reactive groups include, but are not limited to, active esters, active carbonates, aldehydes, isocyanates, isothiocyanates, epoxides, alcohols, maleimides, vinylsulfones, hydrazides, dithiopyridines, and iodoacetamides.

25 The activated water soluble polymer can be covalently linked to a biologically active agent to form a conjugate. A suitable biologically active agent can be any of those having a moiety capable of reacting with at least one of the two reactive groups in the terminus of the activated polymer. The biologically active agent can have two such moieties and each of them can be linked to one of the two
30 reactive groups. Alternatively, the conjugate can have two biologically active agents each being linked to one of the two reactive moieties of the activated polymer. Because activated polymers having different tethering groups can be prepared in accordance with this invention, an activated polymer can be provided

in which the two reactive groups in a terminus of the activated polymer are a desirable distance from each other. When such an activated polymer is conjugated to two biologically active agent molecules, the two molecules can be held at a desired distance apart.

5 Accordingly, the activated PEG can be used with greater versatility as compared to other PEG derivatives heretofore known in the art to form various conjugates with molecules such as proteins or peptides. Since PEG molecules conjugated to another molecule can impart water solubility and reduced immunogenicity to the other molecule, , the activated PEG derivatives of this
10 invention allows greater control and precision in modifying such characteristics in a conjugate.

 Thus, an activated water soluble polymer having proximal reactive groups is provided. The polymer backbone has at least one terminus having two reactive groups. The terminus has a branching moiety and two free reactive
15 moieties linked to the branching moiety. The branching moiety is in turn linked to the polymer backbone through a stable linkage.

DETAILED DESCRIPTION OF THE INVENTION

 The terms "group," "functional group," "moiety," "active moiety," "reactive site," reactive group" and "reactive moiety" are used in the art and herein
20 to refer to distinct, definable portions or units of a molecule. The terms are somewhat synonymous in the chemical arts and are used herein to indicate that the portions of molecules that perform some function or activity and are reactive with other portions of molecules.

 The term "linkage" is used herein to refer to groups or bonds that
25 normally are formed as the result of a chemical reaction and typically are covalent linkages. Hydrolytically stable linkages means that the linkages are substantially stable in water and do not react with water at useful pHs, e.g., under physiological conditions for an extended period of time, preferably indefinitely.

 The term "biologically active agent" when used herein means any
30 substance which can affect any physical or biochemical properties of a biological organism, including but not limited to viruses, bacteria, fungi, plants, animals, and humans. In particular, as used herein, biologically active agent includes any

substance intended for the diagnosis, cure, mitigation, treatment, or prevention of disease in humans or other animals, or to otherwise enhance physical or mental well being of humans or animals. Examples of biologically active agents include, but are not limited to, organic and inorganic compounds, proteins, peptides, lipids, polysaccharides, nucleotides, DNAs, RNAs, other polymers, and derivatives thereof. Examples of biologically active agents include antibiotics, fungicides, anti-viral agents, anti-inflammatory agents, anti-tumor agents, cardiovascular agents, anti-anxiety agents, hormones, growth factors, steroidal agents, and the like. Other examples include, microorganisms such as bacteria and yeast cells, viral particles, plant or animal or human cells, and the like.

The polymer backbone is a water soluble substantially non-immunogenic polymer, and is preferably poly(ethylene glycol) (PEG). However, it should be understood that other related polymers are also suitable for use in the practice of this invention and that the use of the term PEG or poly(ethylene glycol) is intended to be inclusive and not exclusive in this respect.

Poly(ethylene glycol) or PEG is useful in biological applications because it has properties that are highly desirable and is generally approved for biological or biotechnical applications. PEG typically is clear, colorless, odorless, soluble in water, stable to heat, inert to many chemical agents, does not hydrolyze or deteriorate, and is generally nontoxic. Poly(ethylene glycol) is considered to be biocompatible, which is to say that PEG is capable of coexistence with living tissues or organisms without causing harm. More specifically, PEG is non-immunogenic, which is to say that PEG does not tend to produce an immune response in the body. When attached to a molecule having some desirable function in the body, such as a biologically active agent, to form a conjugate, the PEG tends to mask the agent and can reduce or eliminate any immune response so that an organism can tolerate the presence of the agent. Accordingly, the conjugate is substantially non-toxic. PEG conjugates tend not to produce a substantial immune response or cause clotting or other undesirable effects. PEG having the formula - $\text{CH}_2\text{CH}_2-(\text{CH}_2\text{CH}_2\text{O})_n-\text{CH}_2\text{CH}_2-$, where n is from about 8 to about 4000, is one useful polymer in the practice of the invention. Preferably PEG having a molecular weight of from about 200 to about 100,000 Da is used as polymer backbone.

The polymer backbone can be linear or branched. Branched polymer backbones are generally known in the art. Typically, a branched polymer has a central branch core moiety and a plurality of linear polymer chains linked to the central branch core. PEG is commonly used in branched forms that can be prepared by addition of ethylene oxide to various polyols, such as glycerol, pentaerythritol and sorbitol. For example, the four-arm, branched PEG prepared from pentaerythritol is shown below:



10

The central branch moiety can also be derived from several amino acids, e.g., lysine.

The branched polyethylene glycols can be represented in general form as $\text{R}(-\text{PEG-OH})_n$ in which R represents the core moiety, such as glycerol or pentaerythritol, and n represents the number of arms. Suitable branched PEGs can be prepared in accordance with International Publication No. WO 96/21469, entitled *Multi-Armed, Monofunctional, and Hydrolytically Stable Derivatives of Poly(Ethylene Glycol) and Related Polymers For Modification of Surfaces and Molecules*, which was filed January 11, 1996, the contents of which are incorporated herein in their entirety by reference. These branched PEGs can then be modified in accordance with the teachings herein.

Many other water soluble substantially non-immunogenic polymers than PEG are also suitable for the present invention. These other polymers can be either in linear form or branched form, and include, but are not limited to, other poly(alkylene oxides) such as poly(propylene glycol) ("PPG"), copolymers of ethylene glycol and propylene glycol and the like; poly(vinyl alcohol) ("PVA") and the like. The polymers can be homopolymers or random or block copolymers and terpolymers based on the monomers of the above polymers, straight chain or branched.

Specific examples of suitable additional polymers include, but are not limited to, difunctional poly(acryloylmorpholine) ("PACM"), and poly(vinylpyrrolidone) ("PVP"). PVP and poly(oxazoline) are well known polymers in the art and their preparation should be readily apparent to the skilled

artisan. PAcM and its synthesis and use are described in U.S. Patent Nos. 5,629,384 and 5,631,322, the contents of which are incorporated herein by reference in their entirety.

Although the molecular weight of each chain of the polymer backbone can vary, it is typically in the range of from about 100 to about 100,000, preferably from about 6,000 to about 80,000.

Those of ordinary skill in the art will recognize that the foregoing list for substantially water soluble non-immunogenic polymer backbone is by no means exhaustive and is merely illustrative, and that all polymeric materials having the qualities described above are contemplated.

The activated polymer of this invention also has proximal reactive groups linked to at least one arm of the polymer backbone. As will be apparent, the term "proximal" is used herein to mean that the terminus has two free reactive moieties capable of reacting with two other moieties in another molecule or two other molecules, which can be the same or different.

The terminus typically has a branching moiety covalently linked to a polymer chain of the polymer backbone through a hydrolytically stable linkage. Typically, there are two free reactive groups branching out from the branching moiety. The term "free" as used herein means that each of the two free reactive groups has two ends, one of which is covalently linked to the branching moiety and the other end is not linked to any other moiety or group through covalent linkage, and (available for reaction with another moiety or group, e.g., of another molecule).

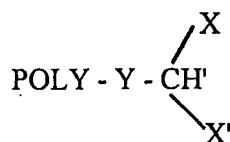
Typically the branching moiety is a stable, non-reactive, and inert moiety that is covalently linked to a polymer chain and to the two reactive groups. The branching moiety should not form a hydrogen bond or ionic bond with metal ions or moieties or molecules. It is believed that the ability to form strong hydrogen bonds or ionic bonds would interfere with the branching moiety's function. The branching atom, i.e., the atom the two free reactive groups are linked to is not a nitrogen atom (N), but is typically a carbon atom (C).

At least one of the two free reactive groups may comprise two portions: a reactive moiety at the free end and a tethering group linking the reactive moiety to the branching moiety. The reactive moiety is a moiety capable of

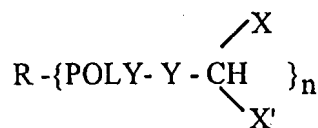
reacting with a moiety in another molecule, e.g., a biologically active agent such as proteins, peptides, etc. Examples of suitable reactive moieties include, but are not limited to, active esters, active carbonates, aldehydes, isocyanates, isothiocyanates, epoxides, alcohols, maleimides, vinylsulfones, hydrazides, dithiopyridines, N-succinimidyl, and iodoacetamides. The selection of a free reactive moiety is determined by the moiety in another molecule to which the free reactive moiety is to react. For example, when the moiety in another molecule is a thiol moiety, then a vinyl sulfone moiety is preferred for the free reactive moiety of the activated polymer. On the other hand, an N-succinimidyl moiety is preferred to react to an amino moiety in a biologically active agent.

The tethering group can have a predetermined length such that the reactive moiety linked to it is at a predetermined distance away from the branching moiety, and consequently, a predetermined distance from the other reactive moiety of the terminus. Typically, the tethering group is non-reactive and is a substantially linear chain of atoms, e.g., alkyl chains, ether chains, ester chains, amide chains, and combinations thereof.

Thus, in a preferred embodiment, the activated polymer of this invention can be represented by formula I:



or formula II:



In the above formula, POLY is a linear polymer chain of a water soluble substantially non-immunogenic polymer backbone, preferably is poly(ethylene glycol) or a derivative thereof. In the activated polymer represented

by formula I, the polymer backbone has only one polymer chain. Y is a hydrolytically stable linkage, which can comprise an atom or a group such as -O-, -S- and -CO-NH-. It will be apparent to skilled artisan that many other hydrolytically stable linkages can also be employed in this embodiment.

5 X and X' are free reactive groups, which can be same or different, each having a reactive moiety capable of reacting with a moiety in another molecule such as a protein. In the activated polymer as represented by formula I, the polymer backbone POLY can have a capping group at the end opposite to the terminus having proximal reactive groups. The capping group can be, for example,
 10 -OH, various alkyl, and can also contain proximal reactive groups -Y-CHXX' wherein Y, X and X' are as described above. Accordingly, the activated polymer can have two terminals with proximal reactive groups, one on each end of the polymer backbone.

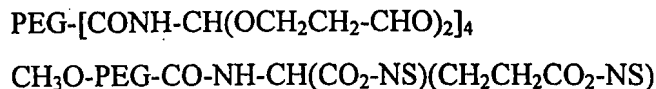
In formula II, R is the central core as described above. POLY is a
 15 polymer chain of the water soluble substantially non-immunogenic polymer backbone. Y is a hydrolytically stable linkage. n is from 2 to 200 representing the number of polymer chains or arms in the polymer backbone, as described above.

As will be apparent, the branching moiety as described above is CH
 in this embodiment of the activated polymer. Typically it does not become
 20 charged in normal conditions, and does not form an ionic bond with a metal ion.

In a preferred embodiment, X and X' can have a tethering group in addition to a reactive moiety and can be represented by -W-Z and -W'-Z' respectively, in which Z and Z' represent free reactive moieties for conjugating the polymer to another molecule. W and W' represent tethering groups. Z and Z' can
 25 be different or same reactive moieties.

Some examples of preferred embodiments of the activated polymers of this invention are provided as follows:

CH₃O-PEG-CO-NH-CH(CH₂-OCO₂-NS)₂
 CH₃O-PEG-CO-NH-CH(CH₂-O₂C-CH₂CH₂-CO₂-NS)₂
 30 CH₃O-PEG-CO-NH-CH(CH₂-CO₂-NS)₂
 CH₃O-PEG-O-CH(CH₂-OCO₂-NS)₂
 CH₃O-PEG-O-CH(CH₂-O₂C-CH₂CH₂-CO₂-NS)₂
 (OHC-CH₂CH₂-O-CH₂)₂-CH-NHCO-PEG-CONH-CH(OCH₂CH₂-CHO)₂



In these examples, -NS represents the N-succinimidyl moiety.

In accordance with another aspect of this invention, a method for
5 preparing the activated water soluble polymer of this invention is also provided.

Typically, in the first step, there is a first intermediate polymer provided having a polymer backbone and a reactive end group covalently linked to of the polymer backbone.

In addition, a compound having three reactive groups linked to a
10 branching moiety is provided. This compound typically has a branching moiety forming a central core and three free groups branching out from the central core. When the three free groups are linked to the same atom in the branching moiety, the atom is not a nitrogen atom. One of the three free groups can react with the reactive end group of the first intermediate polymer to form a hydrolytically stable
15 linkage. The other two free groups can be ultimately converted into the two free reactive groups on the terminus of the activated polymer of this invention.

Examples of these compounds include, $\text{H}_2\text{NCH}(\text{CH}_2-\text{OH})_2$, $\text{NaO}-\text{CH}(\text{CH}_2-\text{O}-\text{Bz})_2$, $\text{H}_2\text{N}-\text{CH}(\text{CH}_2\text{CO}_2\text{H})_2$, and the like. As will be apparent, in these examples, the branching moiety is CH. The $\text{H}_2\text{N}-$ and $\text{NaO}-$ moieties can be used to link the
20 compound to the first intermediate polymer to form a hydrolytically stable linkage, while the hydroxyl groups, carboxylic acid groups, and $-\text{CH}_2-\text{O}-\text{Bz}$ groups can be ultimately converted into free reactive moieties of the activated polymer of this invention.

Thus, in the second step of the method, the compound having three ,
25 reactive groups is reacted with the first intermediate polymer to form a second intermediate polymer which includes a hydrolytically stable linkage linking the first intermediate polymer and the compound having three reactive groups, thus leaving only two free groups at the terminus of the polymer chain.

In the third step, the two free groups of the compound are converted
30 into two free reactive moieties linked to the branching moiety. A number of methods known in the art can be employed in the conversion. For example, the free groups can be reacted to a compound which can impart a free reactive moiety. Alternatively, the two free groups in the second intermediate polymer can be

oxidized or reduced or substituted to form two new free reactive moieties. Such methods will be apparent to skilled artisan in view of the Examples given below.

In accordance with yet another aspect of this invention, a conjugate is provided formed by covalently linking the activated water soluble polymer of this invention to another molecule, e.g., a biologically active agent. Typically, a suitable biologically active agent can be any biologically active agent having a moiety capable of reacting with at least one of the two proximal reactive groups in the terminus of the activated polymer.

The biologically active agent can have two such moieties and each of them can be linked to one of the two reactive groups. Alternatively, the conjugate can have two biologically active agents each being linked to one of the two reactive moieties of the activated polymer. For example, the reactive moieties in the activated polymer can be vinyl sulfone moieties, which can react with a thiol moiety. If a protein has only one thiol moiety, then two of such protein molecules can be linked to the activated polymer through the two vinyl sulfone moieties. When a protein has two thiol moieties, the reaction between the protein and the activated polymer can be controlled such that each activated polymer molecule is conjugated to two protein molecules. Alternatively, the reaction can also be controlled such that the two vinyl sulfone moieties of an activated polymer are reacted with two thiol moieties on the same protein molecule.

Other moieties in biologically active agents useful for reacting with the free reactive moieties of the bivalent terminus of an activated polymer of this invention include, e.g., amino groups, carboxylic acid groups, etc. It will be apparent for skilled artisan once apprised of the present invention to select appropriate free reactive moieties in an activated polymer for reaction with a given moiety in a biologically active agent. For example, if conjugation is through reaction with an amino group in a biologically active agent, moieties such as $-CO_2-$ NS or aldehyde is preferably used as a free reactive moiety in the activated polymer for conjugation.

Because activated polymers having different tethering groups can be prepared in accordance with this invention, an activated polymer can be provided in which the two reactive groups in a bivalent terminus of the activated polymer are in a desirable distance from each other. When such an activated

polymer is conjugated to two biologically active agent molecules, the two molecules can be held at a desired distance apart.

The following examples are given to illustrate the invention, but should not be considered in limitation of the invention:

- 5 Example 1. Synthesis of $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{O}_2\text{CCH}_2\text{CH}_2\text{CO}_2\text{NS})_2$
(NS=N-succinimidyl)

Example 2. Synthesis of $m\text{PEG}_{20K}\text{-OCH}(\text{CH}_2\text{-SO}_2\text{CH=CH}_2)_2$

10

Example 3. Synthesis of $m\text{PEG}_{5K}\text{-O}_2\text{CNH-CH}(\text{CH}_2\text{CO}_2\text{NS})_2$

Example 4. Synthesis of $m\text{PEG}_{5K}\text{-O-CH}_2\text{CH}_2\text{CH}(\text{CO}_2\text{H})_2$

EXAMPLE 1

Reactions:

15

1. $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CO}_2\text{NS} + \text{H}_2\text{NCH}(\text{CH}_2\text{-OH})_2$
 $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{OH})_2 + \text{NHS}$
NS = N-succinimidyl;
NHS = N-hydroxysuccinimide

20

2. $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-OH})_2 + 2\text{SA}$
 $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-O}_2\text{CCH}_2\text{CH}_2\text{CO}_2\text{H})_2$
SA = succinic anhydride

25

3. $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-O}_2\text{CCH}_2\text{CH}_2\text{CO}_2\text{H})_2 + \text{NHS} + \text{DCC}$
 $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-O}_2\text{CCH}_2\text{CH}_2\text{CONS})_2$
DCC = dicyclohexylcarbodiimide

Preparations:

30

1. $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{OH})_2$
A solution of $m\text{PEG}_{20K}\text{-OCH}_2\text{CH}_2\text{CO}_2\text{NS}$ (mSPA 20K, 20 g, 0.001 moles), $\text{H}_2\text{NCH}(\text{CH}_2\text{-OH})_2$ (serinol, 0.14 g, 0.00154 moles), and triethylamine

(0.3 ml) in acetonitrile (100 ml) was stirred under nitrogen overnight and the solvent removed by distillation. The product was chromatographed on DEAE sepharose eluted with water and the eluate was saturated with NaCl and extracted with chloroform. The resulting chloroform phase was dried over magnesium sulfate, filtered, and the filtrate evaporated to dryness under vacuum to yield 20 g of product as a white solid showing a single peak with gel permeation chromatography (Ultrasphere 250, pH 7.2 buffer).

2. $\text{mPEG}_{20\text{K}}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-O}_2\text{CCH}_2\text{CH}_2\text{CO}_2\text{H})_2$

10 A solution of the product from (1) (20 g, .002 moles) and butylated hydroxytoluene (BHT) (0.02g) in 220 ml of chloroform was subjected to distillation until about 150 ml of solvent had distilled. Succinic anhydride (2.0 g, 0.02 moles), pyridine (1.62 ml, 0.02 moles), and 40 ml of toluene were added and the resulting mixture heated at 84 °C for 20 h under nitrogen. The product was
15 precipitated with 850 ml of ether and collected by filtration. After drying, the product was dissolved in 200 ml of water, 20 g of NaCl added, and the pH adjusted to 3 with aqueous phosphoric acid. The product was extracted with chloroform (200 + 150 + 100 ml) and the combined extracts dried over magnesium sulfate. Evaporation of the dried solution yielded the product as a white solid (16 g). The
20 molecular weight was determined to be 20,940 Da by potentiometric titration.

3. $\text{mPEG}_{20\text{K}}\text{-OCH}_2\text{CH}_2\text{CONHCH}(\text{CH}_2\text{-O}_2\text{CCH}_2\text{CH}_2\text{CONS})_2$

A solution of the product from (2) (15 g, 0.0015 moles), N-hydroxysuccinimide (0.21 g, 0.00179 moles), dicyclohexylcarbodiimide, 0.37 g,
25 0.00177 moles) in methylene chloride (100 ml) was stirred at room temperature under nitrogen overnight. The suspension was filtered, product precipitated twice from methylene chloride by addition of ether (850 ml) and collected by filtration to obtain a white solid (13.0 g) which had 97.7% substitution by proton nmr. The proton nmr displayed a broad multiplet at 3.50 ppm (PEG backbone methylene
30 groups), a singlet at 3.23 ppm (PEG methyl), a singlet at 2.80 ppm (NS methylenes), and multiplets at 2.68 and 2.95 ppm (succinate methylenes).

EXAMPLE 2

Reactions:

1. $\text{HO-CH}(\text{CH}_2\text{-O-Bz})_2 + \text{NaH (toluene)} \rightarrow \text{NaO-CH}(\text{CH}_2\text{OBz})_2$
 $\text{Bz} = \text{Benzyl}$
2. $\text{NaO-CH}(\text{CH}_2\text{-O-Bz})_2 + \text{mPEG}_{20\text{K}}\text{-O-Ms} \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OBz})_2$
 $\text{Ms} = \text{mesylate}$
3. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OBz})_2 + \text{HCO}_2\text{H/MeOH/H}_2\text{O/Pd/C} \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OH})_2$
4. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OH})_2 + \text{MsCl/Et}_3\text{N} \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OMs})_2$
5. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OMs})_2 + \text{HSCH}_2\text{CH}_2\text{OH} \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{SCH}_2\text{CH}_2\text{OH})_2$
6. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{SCH}_2\text{CH}_2\text{OH})_2 + \text{H}_2\text{WO}_4 \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{SO}_2\text{CH}_2\text{CH}_2\text{OH})_2$
7. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{SO}_2\text{CH}_2\text{CH}_2\text{OH})_2 + \text{MsCl/ET}_3\text{N} \rightarrow \text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{SO}_2\text{CH=CH}_2)_2$

Preparations:

1. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OBz})_2$
 A solution of 18 g (0,0641 moles) of 1,3-dibenzyloxy-2-propanol in 80 ml of toluene was distilled until 15 ml of toluene was removed. The azeotropically dried solution was then added to a suspension of 2.56 g (0.064 moles) of NaH in 80 ml of toluene and the resulting mixture stirred while heating to 37-40 °C before filtering. The filtrate was then added to a solution of azeotropically-dried mPEG_{20K} mesylate in about 350 ml of toluene and the resulting mixture was heated for 20 h at 125 °C under N₂. The product was precipitated with cold ether, wash on the filter with hexane, and dried under vacuum to yield 70.4 g of white solid shown to be pure by proton nmr.

2. $\text{mPEG}_{20\text{K}}\text{-O-CH}(\text{CH}_2\text{OH})_2$

To a solution of 15 g (0.00075 moles) of the product from (1) in 9.2 ml of formic acid and 0.8 ml of water was added 2.0 g of Pd/C (10%) and the mixture was stirred for 2 h under nitrogen. The mixture was then filtered and the pH of the filtrate adjusted to 7.2. The resulting solution was extracted with CH₂Cl₂ and the extract dried over MgSO₄. Evaporation of the solution yielded 12.9 g of product which displayed no benzyl groups in the proton nmr.

3. mPEG_{20K}O-CH(CH₂OMs)₂

To an azeotropically-dried solution of the product from (2) (8.0 g, 0.000889 moles) in toluene (100 ml) containing 0.008 g of BHT was added a solution of mesyl chloride (0.090 ml, 0.00116 moles) and triethylamine (0.210 ml, 0.0011 moles) in 10 ml of ET₃N and the resulting solution was stirred overnight at room temperature under nitrogen. Ethanol (1 ml) was added and 50 ml of the solvent was removed by distillation before adding 500 ml of ether to precipitate the product. The product was collected by filtration and dried under vacuum to yield 7.6 g of the mesylate derivative shown by nmr to be 100% substituted.

4. mPEG_{20K}O-CH(CH₂SCH₂CH₂OH)₂

A solution of the product of (3), (7.0 g, 0.00035 moles), mercaptoethanol (0.56, 0.0080 moles) ml, NaOH (0.22 g), in toluene (30 ml and ethanol (60 ml) was heated at 60 °C for 2 h under N₂. The pH was adjusted to 7 and the product extracted with methylene chloride (3x100 ml). After drying the extract over MgSO₄, the solvent was removed and the product precipitated with 250 ml of ethyl ether. The product was collected by filtration and dried under vacuum to get 6.6 g of white solid which was shown by nmr to be 97.3% substituted.

5. mPEG_{20K}O-CH(CH₂SO₂CH₂CH₂OH)₂

A solution containing the product from (4), 6.5 g (0.00065 moles), and tungstic acid (0.16 g) in water (14 ml) was prepared and the pH adjusted to 6.6. Hydrogen peroxide (30%, 0.65 ml) was added and the mixture stirred at room temperature overnight. The pH was adjusted to 7.5 and the mixture stirred 1 h before extracting with CH₂Cl₂ (3x30 ml). The mixture was dried over MgSO₄,

filtered, and the filtrate concentrated to 25 ml. The product was precipitated with 200 ml of ether and collected by filtration to obtain 5.3 g of product after vacuum drying. The product was shown by nmr to have 86% substitution.

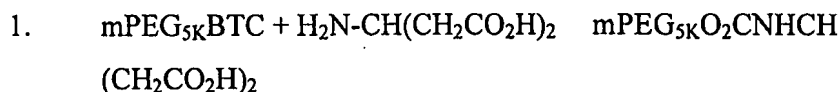
5 6. $m\text{PEG}_{20K}\text{O}-\text{CH}(\text{CH}_2\text{SO}_2\text{CH}=\text{CH}_2)_2$

A solution of the product from (5), (5.2 g, 0.00052 moles), Et_3N (0.63 ml, 0.00452 moles), BHT (0.005 g), and MsCl (0.15 ml, 0.001938 moles) in CH_2Cl_2 (25 ml) was stirred at room temperature for 42 h at room temperature. Ethanol (1 ml) was added and the mixture was stirred 15 minutes. Methylene
 10 chloride (50 ml) was added and the resulting solution was washed with aqueous 1M HCl followed by 5% aqueous Na_2HPO_4 . After drying over MgSO_4 , the solution was concentrated to 30 ml and the product precipitated with 300 ml of ether. The product was collected by filtration and dried under vacuum to yield the product
 15 (4.6 g) as a white solid. The degree of substitution was 92.5% by nmr. The ^1H nmr spectrum ($\text{dmso}-d_6$) displayed absorptions at 3.51 ppm (PEG backbone CH_2), 3.23 ppm, CH_3O), 6.2 and 7.0 ppm, m, vinyl H. Note in this example that $\text{Y} = \text{O}$, $\text{W} = \text{CH}_2$, and $\text{Z} = \text{SO}_2\text{CH}=\text{CH}_2$.

EXAMPLE 3

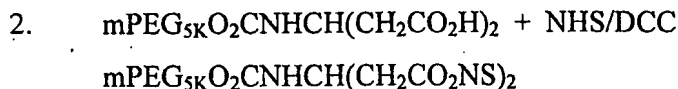
20

Reactions:



25

$\text{BTC} = 1\text{-benzotriazolyl carbonate}$



30 Preparations:



To a solution of β -glutamic acid (0.10 g, 0.00068 moles), boric acid (0.1 g) in 10 ml of water at pH 8 was added $m\text{PEG}_{5K}\text{BTC}$ over 15 m while

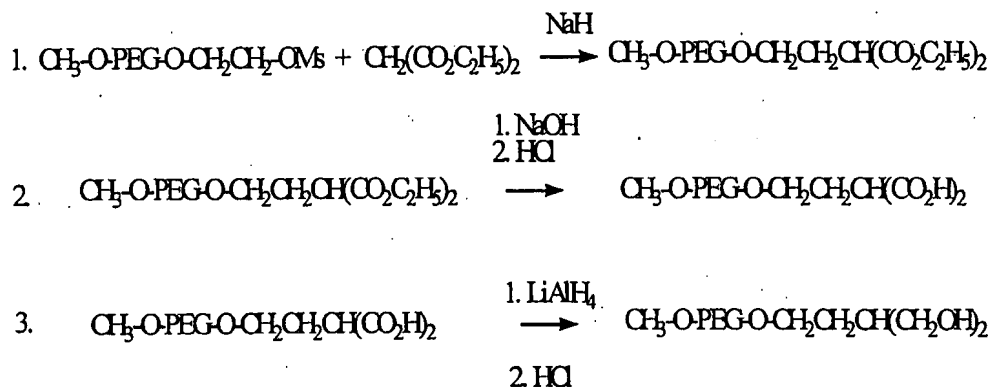
maintaining the pH at 8.15-8.25 by addition of NaOH solution. NaCl (6 g) was added and the pH of the solution was adjusted to 2 with 10% H₃PO₄. The product was extracted into CH₂Cl₂ (100 + 80 + 50 ml) and the combined extracts were dried over MgSO₄. The mixture was filtered and the filtrate evaporated under vacuum to yield 7.8 g of product. The mixture was determined to be 75.5% of the mPEG glutamic acid derivative and 24.5% mPEG. This mixture was purified by chromatography on DEAE sepharose by first eluting with water and then eluting the desired product with 0.5 M NaCl. Extraction of the product from the NaCl solution (pH 2) with methylene chloride followed by drying the extract over MgSO₄ and evaporation of the solvent yielded 6.1 g of material shown to be 100% pure by GPC.

2. $\text{mPEG}_{5K}\text{O}_2\text{CNHCH}(\text{CH}_2\text{CO}_2\text{NS})_2$

A solution of $\text{mPEG}_{5K}\text{O}_2\text{CNHCH}(\text{CH}_2\text{CO}_2\text{H})_2$ (6.0 g, 0.00116 moles), NHS (0.385 g, 0.001627 moles), DCC (0.676 g, 0.00162 moles) in methylene chloride (50 ml) was stirred overnight at room temperature under nitrogen. The resulting suspension was filtered and the filtrate was added to 500 ml of cold ethyl ether. The precipitate was collected by filtration and dried under vacuum to obtain 5.5 g of product which was shown by nmr to have 100% substitution. The ¹H nmr spectrum (dmso-d₆) displayed absorptions at 3.51 ppm (PEG backbone CH₂), 3.23 ppm (CH₃O), 4.29 ppm (-NHCH-), 4.05 ppm (-CH₂-O-CONH-, 3.24 ppm (CH₂CO₂NS), 2.81 (NS CH₂).

EXAMPLE 4

Reactions:



5 Preparations:

1. Preparation of $\text{CH}_3\text{-O-PEG-O-CH}_2\text{CH}_2\text{CH}(\text{CO}_2\text{H})_2$ (Steps 1 and 2 above)

Diethyl malonate (8.8 ml) in 150 ml of dry dioxane was added dropwise to NaH (2.4 g) in 60 ml of toluene under argon. MPEG₅₀₀₀ mesylate (30 g) in 250 ml of toluene was azeotropically distilled to remove 150 ml of toluene and the residue was added to the above diethyl malonate solution. After refluxing the mixture for 3-4 hours, it was evaporated under vacuum to dryness and dried *in vacuo* overnight. The dried material was then dissolved in 200 ml of 1N NaOH, the solution was stirred for 2 days at room temperature, and the pH adjusted to 3 with 1N HCl. NaCl was added to the solution to a concentration of about 15% and the mixture was then extracted with 350 ml of CH_2Cl_2 in several portions. The combined extracts were dried over Na_2SO_4 , concentrated under vacuum and the product precipitated by addition of isopropanol / ether (1:1). The product was collected by filtration and dried under vacuum overnight to obtain 24.7 g of product as a white powder. GPC (Ultrasphere 250) showed the product to be 98% pure.

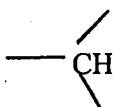
$^1\text{H NMR}$ (dmso- d_6 , ppm): 1.96 (t, $\text{CH}_2\text{CH}_2\text{-C}$); 3.51 (br m, PEG $-\text{CH}_2\text{CH}_2\text{-O-}$).

2. Preparation of $\text{CH}_3\text{-O-PEG}_{5000}\text{-O-CH}_2\text{CH}_2\text{CH}(\text{CH}_2\text{OH})_2$

- 5 $\text{CH}_3\text{-O-PEG}_{5000}\text{-O-CH}_2\text{CH}_2\text{CH}(\text{CO}_2\text{H})_2$ (5 g) was dissolved in 50 ml of toluene and 9.8 ml of LiAlH_4 (1 M in THF) was added. After stirring overnight at room temperature, the mixture was evaporated to dryness under vacuum and 150 ml of water and 22.5 g of NaCl were added. The pH was adjusted to 6.5 with aqueous HCl and the resulting solution was extracted with 3x50 ml of methylene chloride.
- 10 The combined extracts were dried over Na_2SO_4 and the solution was evaporated to dryness. The product was precipitated with ethyl ether and collected by filtration. After chromatography on DEAE sepharose, the product was 90% pure by GPC (Ultrahydrogel 250).
- $^1\text{H NMR}$ (dmso-d_6 , ppm): 3.51 (br m, PEG $-\text{CH}_2\text{CH}_2\text{-O-}$); 1.5 (br mult, CH; 4.32
- 15 (t, OH).

WHAT IS CLAIMED IS:

1. An activated, water soluble polymer comprising a polymer backbone and at least one terminus linked to said polymer backbone through a stable linkage, wherein said at least one terminus comprises a branching moiety and proximal reactive groups linked to said branching moiety that are capable of forming
5 covalent linkages with another molecule.
2. The polymer of Claim 1, wherein at least one of said proximal reactive groups further comprises a tethering group linking said reactive moiety to said branching moiety.
3. The polymer of Claim 2, wherein said tethering group is an alkyl chain.
4. The polymer of Claim 1, wherein said polymer backbone is selected from the group consisting of linear or branched poly(ethylene glycol), linear or branched poly(alkylene oxide), linear or branched poly(vinyl pyrrolidone), linear or branched poly(vinyl alcohol), linear or branched polyoxazoline, and linear or
5 branched poly(acryloylmorpholine).
5. The polymer of Claim 1, wherein said polymer backbone is poly(ethylene glycol).
6. The polymer of Claim 1, wherein said branching moiety is



7. An activated water soluble polymer having the formula:
POLY - Y - CHXX'
wherein POLY is a polymer backbone, Y is a hydrolytically stable linkage, and X and X' are reactive groups having a reactive moiety, which may be the same or
5 different, and are selected from the group consisting of active esters, active carbonates, aldehydes, isocyanates, isothiocyanates, epoxides, alcohols, maleimides, vinylsulfones, hydrazides, dithiopyridines, and iodoacetamides.

8. The polymer of Claim 7, wherein POLY has a capping group on the opposite terminus selected from the group consisting of -OH, alkyl, and -Y-CHXX'.

9. The polymer of Claim 8, wherein said polymer backbone is selected from the group consisting of linear or branched poly(ethylene glycol), linear or branched poly(alkylene oxide), linear or branched poly(vinyl pyrrolidone), linear or branched poly(vinyl alcohol), linear or branched polyoxazoline, and linear or
5 branched poly(acryloylmorpholine).

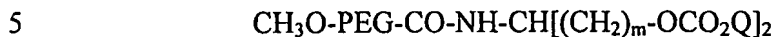
10. The polymer of Claim 8, wherein POLY is poly(ethylene glycol) or derivatives thereof.

11. The polymer of Claim 8, wherein Y is selected from the group consisting of -O-, -S- and -CO-NH-.

12. The polymer of Claim 8, wherein X and X' are represented by -W-Z and -W'-Z' respectively wherein Z and Z' represent said reactive moieties, and W and W' represent a tethering group comprising a substantially linear chain of atoms.

13. The polymer of Claim 12, wherein W and W' are selected from the group consisting of alkyl chains, ether chains, ester chains, amide chains, and combinations thereof.

14. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -CO-NH-, W and W' are -(CH₂)_m - O- where m is 1-10, and Z and Z' are -CO₂Q where Q is N-succinimidyl, sulfo-N-succinimidyl, or the 1-benzotriazolyl group:



15. The polymer of Claim 12, wherein POLY is methoxy-poly(ethylene glycol), Y is -CO-NH-, W and W' are $-(CH_2)_m-O_2C-CH_2CH_2-$ where m is 1-10, and Z and Z' = $-CO_2Q$ where Q is N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group:



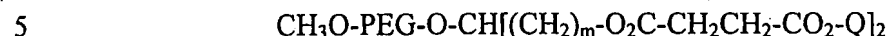
16. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -CO-NH-, W and W' are $-(CH_2)_m-$ where m is 1-10, and Z and Z' are $-CO_2-Q$ where Q is N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group:



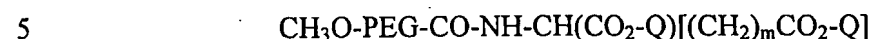
17. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -O-, W and W' are $-(CH_2)_m-O-$ where m is 1-10, and Z and Z' are $-CO_2-Q$ where Q is N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group:



18. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -O-, W and W' are $-(CH_2)_m-O_2C-CH_2CH_2-$ where m is 1-10, and Z and Z' are $-CO_2-Q$ where Q is N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group:



19. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -CO-NH-, and W is a covalent bond directly to -CH-, W' is $(CH_2)_m-$, where m is 1-10, and Z and Z' are $-CO_2-Q$ where Q is an N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group:



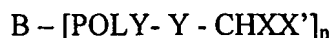
20. The polymer of Claim 12, wherein POLY is methoxy poly(ethylene glycol), Y is -O-, W and W' are $-(CH_2)_m$ where m is 1-10, and Z and Z' are $-SO_2-CH=CH_2$:



21. The polymer of Claim 12, wherein POLY is linear, difunctional poly(ethylene glycol), Y is -CONH-, W and W' are $-(CH_2)_rO(CH_2)_m-$ where r is 1-10 and m is 1-10, and Z and Z' are -CHO:



22. An activated water soluble polymer having the formula:



wherein

- 5 B is a central core;
 POLY is a polymer backbone;
 Y is a hydrolytically stable linkage;
 n is from 2 to 200;
 X and X' are reactive groups having a reactive moiety selected from the
 10 group consisting of active esters, active carbonates, aldehydes, isocyanates, isothiocyanates, epoxides, alcohols, maleimides, vinylsulfones, hydrazides, dithiopyridines, and iodoacetamides.

23. The polymer of Claim 22, wherein POLY is poly(ethylene glycol) or derivatives thereof.

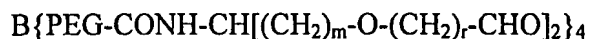
24. The polymer of Claim 22, wherein B is moiety selected from the group consisting of lysine, glycerol, pentaerythritol, and sorbitol.

25. The polymer of Claim 22, wherein Y is selected from the group consisting of -O-, -S- and -CO-NH-.

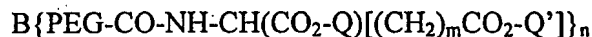
26. The polymer of Claim 22, wherein X and X' are represented by -W-Z and -W'-Z' respectively where Z and Z' represents said reactive moieties, W and W' represent a tethering group comprising a substantially linear chain of atoms.

27. The polymer of Claim 22, wherein W and W' are selected from the group consisting of alkyl chains, ether chains, ester chains, amide chains, and combinations thereof.

28. The polymer of Claim 26, wherein POLY is a poly(ethylene glycol), n is 4, Y is -CONH-, W and W' are $-(CH_2)_r-O(CH_2)_m-$ where r is 1-10 and m is 1-10, and Z and Z' are -CHO:



29. The polymer of Claim 26, the polymer has the formula:



wherein PEG represents poly(ethylene glycol), Y is -CO-NH-, m is 1-10, Q and Q' are N-succinimidyl, sulfo-N-succinimidyl, or 1-benzotriazolyl group, and B and n are as described.

30. A method for preparing an activated water soluble polymer, said polymer comprising of a polymer backbone and at least one branched terminus having proximal reactive groups thereon, said terminus linked to said polymer backbone through a stable linkage, said method comprising:

5 providing a first intermediate polymer comprising said polymer backbone linked to a reactive end group;

providing a reactive compound having three free reactive groups linked to a branching moiety, one of said three free groups being capable of reacting with said reactive end group of the first intermediate polymer to form said stable linkage;

10 and

reacting the reactive end group of said intermediate polymer with said one of said three free groups of the compound to form said stable linkage.

31. The method of Claim 30, wherein at least one of said two free reactive moieties further comprises a tethering group linking said reactive group to said branching moiety.

32. The method of Claim 30, wherein said tethering group is an alkyl chain.

33. The method of Claim 30, wherein said polymer backbone is selected from the group consisting of linear or branched poly(ethylene glycol), linear or branched poly(alkylene oxide), linear or branched poly(vinyl pyrrolidone), linear

or branched poly(vinyl alcohol), linear or branched polyoxazoline, and linear or
 5 branched poly(acryloylmorpholine).

34. The method of Claim 30, wherein said polymer backbone is
 poly(ethylene glycol).

35. The method of Claim 30, wherein said branching moiety is $-\text{CH}-$.

36. The method of Claim 30, wherein said compound is $\text{H}_2\text{NCH}(\text{CH}_2-\text{OH})_2$.

37. The method of Claim 30, wherein said compound is
 $\text{NaO}-\text{CH}(\text{CH}_2-\text{O}-\text{Bz})_2$.

38. The method of Claim 30, wherein said compound is
 $\text{H}_2\text{N}-\text{CH}(\text{CH}_2\text{CO}_2\text{H})_2$.

39. A method for preparing an activated water soluble polymer
 represented by $\text{mPEGO}-\text{CH}(\text{CH}_2\text{SO}_2\text{CH}=\text{CH}_2)_2$, where mPEG represents methyl
 poly(ethylene glycol), comprising the steps of:

- providing mPEG-O-Ms, wherein Ms represents mesylate;
- 5 providing $\text{NaO}-\text{CH}(\text{CH}_2-\text{O}-\text{Bz})_2$, wherein Bz represents a benzyl moiety;
- reacting mPEG-O-Ms with $\text{NaO}-\text{CH}(\text{CH}_2-\text{O}-\text{Bz})_2$ to form
 $\text{mPEGO}-\text{CH}(\text{CH}_2\text{OBz})_2$;
- converting $\text{mPEGO}-\text{CH}(\text{CH}_2\text{OBz})_2$ to $\text{mPEGO}-\text{CH}(\text{CH}_2\text{OH})_2$;
- substituting a mesylate group for both hydroxyl groups in $\text{mPEGO}-$
 10 $\text{CH}(\text{CH}_2\text{OH})_2$ to form $\text{mPEGO}-\text{CH}(\text{CH}_2\text{OMs})_2$;
- converting $\text{mPEGO}-\text{CH}(\text{CH}_2\text{OMs})_2$ into $\text{mPEGO}-\text{CH}(\text{CH}_2\text{SCH}_2\text{CH}_2\text{OH})_2$;
- oxidizing the sulfur atom into a sulfone group; and
- converting the sulfone alcohol group into a vinyl sulfone group, thus
 forming $\text{mPEGO}-\text{CH}(\text{CH}_2\text{SO}_2\text{CH}=\text{CH}_2)_2$.

40. A conjugate comprising:
a biologically active agent; and
an activated water soluble polymer comprising a polymer backbone having
at least one terminus linked to said polymer backbone through a stable linkage,
5 wherein said at least one terminus comprises a branching moiety having proximal
reactive groups linked to said branching moiety,
said biologically active agent being covalently linked to at least one of said
proximal reactive groups.

41. The conjugate of Claim 40, wherein said reactive groups comprise a
reactive moiety capable of reacting with amino groups and said biologically active
agent is selected from the group consisting of proteins, peptides, aminolipids,
polysaccharides having an amino group, amino-oligonucleotides, and
5 pharmaceutical agents having an amino group.

42. The conjugate of Claim 40, wherein said reactive groups comprise
vinyl sulfone and said biologically active agent has a thiol moiety.

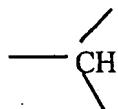
43. The conjugate of Claim 40, wherein at least one of said two reactive
groups comprises a tethering group linking the reactive moiety to said branching
moiety.

44. The conjugate of Claim 43, wherein said tethering group is an alkyl
chain.

45. The conjugate of Claim 40, wherein said polymer backbone is selected
from the group consisting of linear or branched poly(ethylene glycol), linear or
branched poly(alkylene oxide), linear or branched poly(vinyl pyrrolidone), linear
or branched poly(vinyl alcohol), linear or branched polyoxazoline, and linear or
5 branched poly(acryloylmorpholine).

46. The conjugate of Claim 40, wherein said polymer backbone is
poly(ethylene glycol).

47. The conjugate of Claim 40, wherein said branching moiety is



48. The conjugate of Claim 40, wherein said conjugate comprises two separate biologically active agents, which can be the same or different, each being linked to one of said two reactive groups.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 99/05333

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 A61K47/48 C08G65/32

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A61K C08G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	J. M. HARRIS: "LABORATORY SYNTHESIS OF POLYETHYLENE GLYCOL DERIVATIVES" JOURNAL OF MACROMOLECULAR SCIENCE-REVIEWS IN MACROMOLECULAR CHEMISTRY, vol. C-25, no. 3, 1 January 1985, pages 325-373, XP002036446 see the whole document ---	1-13
Y	WO 97 03106 A (SHEARWATER POLYMERS ET AL.) 30 January 1997 see the whole document ---	1-48
Y	WO 96 21469 A (SHEARWATER POLYMERS ET AL.) 18 July 1996 cited in the application see claims --- -/-	1-48

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "3" document member of the same patent family

Date of the actual completion of the international search

15 June 1999

Date of mailing of the international search report

23/06/1999

Name and mailing address of the ISA
European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Boeker, R

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 99/05333

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9703106 A	30-01-1997	US 5672662 A AU 6345796 A	30-09-1997 10-02-1997
WO 9621469 A	18-07-1996	AU 4755596 A	31-07-1996
EP 0839849 A	06-05-1998	JP 10139877 A NO 972846 A US 5767284 A	26-05-1998 06-05-1998 16-06-1998
US 5739208 A	14-04-1998	US 5446090 A US 5900461 A AU 687937 B AU 1054895 A BG 100568 A BR 9408048 A CA 2176203 A CN 1137280 A CZ 9601375 A EP 0728155 A FI 962004 A JP 9508926 T NO 961918 A NZ 276313 A PL 314298 A SK 60896 A WO 9513312 A	29-08-1995 04-05-1999 05-03-1998 29-05-1995 31-12-1996 24-12-1996 18-05-1995 04-12-1996 16-10-1996 28-08-1996 10-05-1996 09-09-1997 12-07-1996 24-04-1997 02-09-1996 09-04-1997 18-05-1995
EP 0605963 A	13-07-1994	AU 5238393 A FI 935485 A JP 7196925 A NO 934477 A NZ 250375 A CA 2110543 A ZA 9309214 A	23-06-1994 10-06-1994 01-08-1995 10-06-1994 26-07-1995 10-06-1994 08-06-1995